סיכום כל הקשרים

$$\begin{tikzcd}
K_a \arrow[r, leftrightarrow, "\frac{10^{-14}}{K}"] \arrow[d, "-\log K"'] & K_b \arrow[d, "-\log K"] \\
pK_a \arrow[r, leftrightarrow, "14-pK"'] \arrow[u, "10^{-pK}"'] & pK_b \arrow[u, "10^{-pK}"]
\end{tikzcd}$$

$$\begin{tikzcd}
[H^+] \arrow[r, leftrightarrow, "\frac{10^{-14}}{[\square]}"] \arrow[d, "-\log[\square]"'] & [OH^-] \arrow[d, "-\log[\square]"] \\
pH \arrow[r, leftrightarrow, "14-p"'] \arrow[u, "10^{-p}"'] & pOH \arrow[u, "10^{-p}"]
\end{tikzcd}$$